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## **Note**

## **Influence of the chain length of chemically bonded phases on the behaviour of several thiazide, potassium-sparing and loop diuretics in high-performance liquid chromatography**

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Thiazide, potassium-sparing and loop diuretics have been analysed by different techniques: paper and thin-layer chromatography<sup>1,2</sup> and high-performance liquid chromatography  $(HPLC)^{3-6}$ . Most of these publications considered only some diuretics. We have investigated both thiazide, potassium-sparing and loop diuretics on a reversed-phase (RP) LiChrosorb  $C_{18}$  column<sup>7</sup>. In order to investigate the selectivity and speed of analysis, other stationary phases have now been used: LiChrosorb RP  $C_8$  and  $C_2$ .

EXPERIMENTAL

#### *Apparatus*

A SP *8770* isocratic pump (Spectra-Physics, Darmstadt, F.R.G.) was equipped with an HP 1040A *W* spectrophotometric detector (Hewlett-Packard, Palo Alto, CA, U.S.A.), an HP 85 computer, an HP 82901M flexible disc drive and an HP 3390A integrator. The detector contained a photodiode array.

The eluent was filtered through a 5- $\mu$ m filter and degassed with helium. A Valco six-port injection valve with a  $10-\mu$  sample loop was used. The column was thermostatted with a water-bath.

## *Chromatographic procedure*

LiChrosorb RP  $C_{18}$ ,  $C_8$  and  $C_2$  columns (Chrompack, Middelburg, The Netherlands) of the same dimensions were used. The chromatographic conditions listed in Table I were used, unless specified otherwise. The organic solvent-water ratios of the eluents are given as volume ratios. The acetonitrile-buffer eluents were prepared by mixing the stated volume percentages. The molarity and pH of the buffer refer to the water phase.

## *Chemicals and reagents*

All reagents were of analytical grade quality; acetonitrile was of HPLC grade quality. The diuretics are listed in Table II. The chemical structures of these diuretics are given in ref. 7.



## TABLE I

## HPLC CONDITIONS

## RESULTS AND DISCUSSION

Thiazide, potassium-sparing and loop diuretics have been analysed on a Li-Chrosorb RP  $C_{18}$  stationary phase<sup>7</sup>. An acetonitrile-water (40:60) eluent was used to separate the thiazide diuretics. The loop and the potassium-sparing diuretics were separated with acetonitrile-phosphate buffer (pH 3) (40:60) as eluent.

Possible changes in the selectivity and elution characteristics were tested on other alkyl stationary phases. The retention characteristics of the diuretics were tested on commercially available LiChrosorb columns: RP  $C_{18}$ ,  $C_8$  and  $C_2$ . Some authors have already dealt with this problem<sup>8-10</sup>. Berendsen and De Galan<sup>8</sup> found that both the capacity factor and the selectivity increase up to a certain alkyl chain length, the

# TABLE II

## DIURETICS STUDIED



 $*$  The numbers of the diuretics are indicated in the text in parentheses.

critical chain length, after which they remain constant. Engelhardt and Ahr<sup>9</sup> mentioned that the increase in selectivity was observed when the stationary phase was wetted by the eluent. However, we observed a decrease in  $k'$  for nearly all thiazide diuretics with increasing chain length (Fig. 1). The  $k'$  values of the acidic and neutral potassium-sparing and loop diuretics showed a slight increase, except for furosemide (14) (Fig. 2). We observed a marked increase in retention only for the basic drugs amiloride hydrochloride (18) and triamterene (17).

This abnormal behaviour of the thiazide diuretics prompted us to check the properties of the commercially available LiChrosorb RP materials. LiChrosorb RP  $C_{18}$  and  $C_8$  are made of 100- $\AA$  silica gel, whereas LiChrosorb RP  $C_2$  is made of 60- $\AA$ silica gel. The specific surface areas of these packings amounted to 125, 250 and 350  $m^2/g$  respectively<sup>11</sup>. The higher specific surface area of the RP C<sub>2</sub> phase might explain this abnormal behaviour. Goldberg<sup>10</sup> also investigated different commercially available RP stationary phases, among them LiChrosorb RP  $C_{18}$ ,  $C_8$  and  $C_2$ . The *k'* values of compounds of a wide polarity range, containing neutral, basic and acidic compounds, were compared: the values increased with increasing chain length; however, the increase observed was much smaller than the increase found by Berendsen and De Galan. Those workers used laboratory-prepared bonded phases on  $10$ -um, 100-A silica particles, whereas Goldberg used commercially available phases. However, none of the compounds tested by Goldberg showed a "decrease" in *k'.* Probably the greater accessibility of the silanol groups on the RP  $C_2$  phase plays an important rôle in this abnormal retention behaviour of the thiazide diuretics. The importance of the "accessibility" of the silanol groups has been discussed previously<sup>12,13</sup>. Bidlingmeyer *et al.*<sup>12</sup> found that the "accessibility" of the silanol groups plays an im-



Fig. 1. Capacity factor of some thiazide diuretics as a function of RP chain length. Eluent: acetonitrilewater (40:60). Chromatographic conditions as in Table I. Key to diuretics as in Table II.

![](_page_3_Figure_1.jpeg)

Fig. 2. Capacity factor of some potassium-sparing and loop diuretics as a function of RP chain length. Eluent: acetonitrile--0.05  $M$  phosphate buffer (pH 3) (40:60). Other details as in Fig. 1.

portant r81e in the peak symmetry, and not the presence or absence of silanol groups. Kohler and Kirkland<sup>13</sup> found that not only the concentration of residual silanol groups, but also their type and availability largely determine the behaviour of bonded-phase columns.

When the stationary phase is changed, only small differences in selectivity are observed for the thiazide diuretics that have an analogous structure (Fig. 1) but a reverse elution order is observed for cyclopenthiazide (12). On the other hand, with the potassium-sparing and loop diuretics that do have different chemical structures, great differences in selectivity are observed when the stationary phase is changed (Fig. 2).

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